## **CLAIM LISTING**

- 1. (PREVIOUSLY PRESENTED) An electrochemical cell comprising:
  - (a) a cathode comprising an electroactive sulfur-containing material;
  - (b) an anode comprising lithium; and
  - (c) a nonaqueous electrolyte, wherein the electrolyte comprises:
    - (i) one or more nonaqueous solvents selected from the group consisting of acyclic ethers, cyclic ethers, polyethers, and sulfones;
    - (ii) one or more lithium salts; and
    - (iii) one or more N-O additives selected from one or more of the group consisting of organic nitrates, inorganic nitrites, potassium nitrate, cesium nitrate, barium nitrate, and ammonium nitrate, nitropropane, nitrobenzene, dinitrobenzene, nitrotoluene, dinitrotoluene, nitropyridine, nitrobutanes, and dialkyl imiazolium.
- (PREVIOUSLY PRESENTED) The cell of claim 1 wherein the one or more N-O additives is selected from one or more of the group consisting of organic nitrates, and inorganic nitrites.
- 3. (PREVIOUSLY PRESENTED) The cell of claim 1 wherein the N-O additive is selected from one or more of the group consisting of potassium nitrate, cesium nitrate, barium nitrate, and ammonium nitrate.
- 4. (ORIGINAL) The cell of claim 2 wherein the organic nitrate is selected from one or more of the group consisting of guanidine nitrate, imidazolium nitrate, and pyridinium nitrate.
- 5. (ORIGINAL) The cell of claim 2 wherein the inorganic nitrite is selected from one or more of the group consisting of lithium nitrite, potassium nitrite, cesium nitrite, and ammonium nitrite.

- 6. (CANCELED)
- 7. (ORIGINAL) The cell of claim 1 wherein the one or more lithium salts is selected from one or more of the group consisting of LiSCN, LiCF<sub>3</sub>SO<sub>3</sub>, and LiN(CF<sub>3</sub>SO<sub>2</sub>)<sub>2</sub>.
- 8. (ORIGINAL) The cell of claim 1 wherein the one or more lithium salts consist of LiSCN and LiN(CF<sub>3</sub>SO<sub>2</sub>)<sub>2</sub> and the N-O additive comprises lithium nitrate.
- 9. (PREVIOUSLY PRESENTED) The cell of claim 1 wherein the concentration of the one or more N-O additives in the electrolyte is from about 0.02 moles/kg of solvent to 2.0 moles/kg of solvent.
- 10. (PREVIOUSLY PRESENTED) The cell of claim 1 wherein the concentration of the one or more N-O additives in the electrolyte is from about 0.1 moles/kg of solvent to 1.5 moles/kg of solvent.
- 11. (PREVIOUSLY PRESENTED) The cell of claim 1 wherein the concentration of the one or more N-O additives in the electrolyte is from about 0.2 moles/kg of solvent to 1.0 moles/kg of solvent.
- 12. (PREVIOUSLY PRESENTED) The cell of claim 1 wherein the concentration of the one or more lithium salts in the electrolyte is from about 0.2 moles/kg of solvent to about 2.0 moles/kg of solvent.
- 13. (ORIGINAL) The cell of claim 1 wherein the acyclic ether is selected from one or more of the group consisting of dimethoxymethane, trimethoxymethane, dimethoxyethane, diethoxyethane, and 1,3-dimethoxypropane.
- 14. (ORIGINAL) The cell of claim 1 wherein the cyclic ether is selected from one or more of the group consisting of tetrahydrofuran, 2-methyl tetrahydrofuran, tetrahydropyran, 1,3-dioxolane, 1,3-dioxane, and 1,4-dioxane.

- 15. (ORIGINAL) The cell of claim 1 wherein the polyether is selected from one or more of the group consisting of diethylene glycol dimethyl ether, triethylene glycol dimethyl ether, tetraethylene glycol dimethyl ether, and dipropylene glycol dimethyl ether.
- 16. (ORIGINAL) The cell of claim 1 wherein the sulfone is selected from one or more of the group consisting of sulfolane, 3-methyl sulfolane, and 3-sulfolene.
- 17. (ORIGINAL) The cell of claim 1 wherein the electroactive sulfur-containing material comprises greater than 75 % by weight of sulfur.
- 18. (ORIGINAL) The cell of claim 1 wherein the electroactive sulfur-containing material comprises elemental sulfur.
- 19. (ORIGINAL) The cell of claim 1 wherein the anode comprises lithium metal.
- 20. (ORIGINAL) The cell of claim 1 that further includes a separator disposed between the anode and the cathode.
- 21. (ORIGINAL) A battery comprising a casing and one or more cells of claim 1.
- 22. (PREVIOUSLY PRESENTED) The cell of claim 1 further comprising lithium nitrate.
- 23. (ORIGINAL) The cell of claim 1 wherein the nonaqueous solvent comprises dioxolane.
- 24. (ORIGINAL) The cell of claim 1 wherein the one or more solvents consists of dimethoxyethane and dioxolane.

- 25. (ORIGINAL) The cell of claim 20 wherein the one or more N-O additives was included as part of the separator and was introduced into the electrolyte after the electrolyte came into contact with the separator.
- 26. (ORIGINAL) The cell of claim 1 wherein the one or more N-O additives was included as part of the cathode and was introduced into the electrolyte after the electrolyte came into contact with the cathode.
- 27. (PREVIOUSLY PRESENTED) An electrochemical cell comprising:
  - (a) a cathode comprising an electroactive sulfur-containing material;
  - (b) an anode comprising lithium; and
  - (c) a nonaqueous electrolyte, wherein the electrolyte comprises:
    - (i) one or more nonaqueous solvents selected from the group consisting of acyclic ethers, cyclic ethers, polyethers, and sulfones; and
  - (ii) one or more N-O additives selected from one or more of the group consisting of organic nitrates, inorganic nitrites, potassium nitrate, cesium nitrate, barium nitrate, and ammonium nitrate.
- 28. (PREVIOUSLY PRESENTED) The cell of claim 27 wherein the one or more N-O additives is selected from one or more of the group consisting of organic nitrates, and inorganic nitrites.
- 29. (PREVIOUSLY PRESENTED) The cell of claim 27 wherein the N-O additive is selected from one or more of the group consisting of potassium nitrate, cesium nitrate, barium nitrate, and ammonium nitrate.
- 30. (ORIGINAL) The cell of claim 29 wherein the inorganic nitrite is selected from one or more of the group consisting of lithium nitrite, potassium nitrite, cesium nitrite, and ammonium nitrite.

- 31. (PREVIOUSLY PRESENTED) The cell of claim 27 wherein the concentration of the one or more N-O additives in the electrolyte is from about 0.2 moles/kg of solvent to 2.0 moles/kg of solvent.
- 32. (ORIGINAL) The cell of claim 27 wherein the electrolyte further comprises one or more lithium salts selected from one or more of the group consisting of LiSCN, LiCF<sub>3</sub>SO<sub>3</sub>, and LiN(CF<sub>3</sub>SO<sub>2</sub>)<sub>2</sub>.
- 33. (ORIGINAL) The cell of claim 27 that further includes a separator disposed between the anode and the cathode.
- 34. (ORIGINAL) The cell of claim 33 wherein the one or more N-O additives was included as part of the separator and introduced into the electrolyte after the electrolyte came into contact with the separator.
- 35. (ORIGINAL) The cell of claim 27 wherein the one or more N-O additives was included as part of the cathode and introduced into the electrolyte after the electrolyte came into contact with the cathode.
- 36. (ORIGINAL) A battery comprising a casing and one or more cells of claim 27.